

## Study of interaction of spironolactone with hydroxypropyl- $\beta$ -cyclodextrin in aqueous solution and in solid state

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### Abstract

The interaction between spironolactone (Sp) and hydroxypropyl- $\beta$ -cyclodextrin (HP $\beta$ CD) has been investigated in aqueous solution and in the solid state. Phase solubility studies indicate that the complex is formed with possible stoichiometry of 1:1 and association constant of  $9.8 \times 10^4 \text{ M}^{-1}$ . Differential scanning calorimetry (DSC) gave evidence for the formation of the complex. The details of Sp/HP $\beta$ CD molecular interaction were analyzed by Fourier transform infrared (FTIR) and Raman spectroscopy. The characterization studies confirmed that the C=O groups of the six-member ring and acetylthio group of Sp interacts with the hydroxyl groups inside and outside the cavity of HP $\beta$ CD through formation of hydrogen bonds. © 2020 Elsevier B.V. All rights reserved.

### Reaxys Database Information

### Author keywords

Complex formation; FT Raman spectroscopy; FTIR spectroscopy; Hydroxypropyl- $\beta$ -cyclodextrin; Spironolactone

### Indexed Keywords

**Engineering controlled terms:** Differential scanning calorimetry; Fourier transform infrared spectroscopy; Hydrogen bonds; Molecular interactions; Raman spectroscopy; Solid state reactions

**Engineering uncontrolled terms:** Acetylthio groups; Complex formation; Hydroxypropyl- $\beta$ -cyclodextrin; Spironolactones

**Engineering main heading:** Ketones

ISSN: 00222860 CODEN: JMOSB Source Type: Journal Original language: English

DOI: 10.1016/j.molstruc.2020.07.043 Document Type: Article